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Chapter Assessment, p. 26 ... observe that indicates a change in matter is occurring. Answers might include a change in appearance or odor produced, or gas and energy produced. •

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Use the odor produced by this ... Chapter 1 • Introduction to Chemistry 3.

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What is a physical change in matter? Molecules can move from one physical state to another (phase change) and not change their atomic structure Chemistry matter and change chapter 7 assessment answers. Oxygen ( $O_2$ ) gas has the same chemical properties as liquid oxygen. Chemistry matter and change chapter 7 assessment answers

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and shapes, and that the differing properties of atoms are due to the size, shape, and movement of the atoms. The Structure of the Atom  
The Structure of the Atom

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided.

1. Classify each of the following as a homogeneous or heterogeneous substance.  
homogeneous a. sugar homogeneous

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d. plastic wrap homogeneous b. iron filings heterogeneous e. cement sidewalk heterogeneous c. granola bar 2. For each type of investigation, select the most appropriate ...

**1 Matter and Change - HUBBARD'S CHEMISTRY - Home Assessment KEYS** 1) What is matter? Matter is something that has mass and volume. 2) Name the three phases (states) of matter? The three most common phases of matter on Earth are solids, liquids and gases. 3) Identify and describe the particle in each phase of matter and how they are different in each phase of matter.

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## **Chapter 3 Assessment Chemistry Matter And Change Answers**

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### **Chapter 7 Assessment Answers Chemistry Matter And Change**

Solutions Manual Chemistry: Matter and Change • Chapter 7 103  
Section 7.1 Ion Formation pages 206–209 Section 7.1  
Assessment page 209 1. Compare the stability of a lithium atom with that of its ion, Li . The Li<sup>+</sup> ion is more stable because it has a complete octet. 2. Describe two different causes of the force of attraction in a chemical bond.

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### **Ionic Compounds and Metals**

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