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PLAY. The basis of an  
ionic bond is the\_\_  
absorption of water

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into their solid  
structures. Elements  
tend to gain or lose  
electrons in order to  
acquire the electron  
configuration of a\_\_  
noble gas. Ions that are  
formed from more than  
one atom are called\_\_

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Chapter 7. monatomic  
ion. binary compounds.  
nomenclature.

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oxyanions. an ion  
formed from a single  
atom. (remember to  
change nonmetal....

compounds composed  
of two different  
elements; (compounds  
of 2.... naming system.  
polyatomic ions that  
contain oxygen.

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Original Test B  
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Answer Key  
PLAY. Match. Gravity.

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Chemical Names &  
Formulas, Oxidation  
Numbers, Using  
Chemical Formulas, &  
Determining Chemical  
Formulas Terms. Terms  
in this set (32)

monatomic ion. an ion  
formed from a single  
atom. (remember to  
change nonmetals to  
-ide ending)

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## Assessment Chapter 7 Test B Flashcards | Quizlet

Modern Chemistry  
Chapter 7 Section 1  
Review - Quizlet -  
many main-group  
elements can lose or  
gain electrons to form  
ions. - ions formed  
from a single atom are  
known as monatomic  
ions. - example: to gain  
a noble-gas electron  
configuration, nitrogen  
gains three electrons  
to form  $N^{3-}$  ions.

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...

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capably as search for  
them. In some cases,  
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## **[Book] Modern Chemistry Chapter 7 Test**

1. Modern Chemistry  
Chapter 7 Chemical  
Formulas. Chemical  
Compounds. A  
chemical formula

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Chapter Test B  
Answer Key

indicates the kind and relative number of atoms in a chemical compound.  $C_8H_{18}$  (octane) has 8 carbon and 18 hydrogen atoms. 2. Forming Ionic Compounds.

## **PPT - Modern Chemistry Chapter 7 Chemical Formulas**

...

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Chapter 7 Section 2  
Review. STUDY.  
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PLAY. Match. Gravity.  
Created by.  
dolphinking. the  
second thing  
merbobber hi hullo  
friends. Terms in this  
set (9) oxidation  
numbers rules. 1. the  
atoms in a pure  
element must equal  
zero

## **Modern Chemistry Chapter 7 Section 2 Review Flashcards**

...

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Assessment  
Modern Chemistry 35

Chapter Test Name B

Class Date Chapter

Test B, continued Write

the orbital notation for  
the following elements  
in the space provided

Modern chemistry

chapter 7 test answer

key. 37. lithium, atomic

number 3 38. carbon,

atomic number 6 39

Modern chemistry

chapter 7 test answer

key. neon, atomic

number 10 PART VI

Write the answers to



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Assessment  
the questions on the  
line to the left, and B.

## Chapter Test B Answer Key **Modern Chemistry Chapter 7 Test Answer Key**

Holt McDougal Modern  
Chemistry 1 Chapter  
Test Assessment

Chapter Test A

Chapter: The Periodic  
Law Use the periodic  
table below to answer  
the questions in this  
Chapter Test. In the  
space provided, write  
the letter of the term

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Assessment  
Chapter Test B  
Answer Key

or phrase that best  
completes each  
statement or best  
answers each question.

## **Holt Modern Chemistry Chapter 7 Test Answers**

Modern Chemistry 141

Chapter Test Name

Class Date Chapter

Test B, continued 18.

When a strong acid is

titrated with a weak

base, the pH of the

solution at the

equivalence point is

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Assessment  
Chapter Test B  
Answer Key

than 7. 19. A is a highly purified solid used to check the concentration of a standard solution. 20. A 1 M solution of NaOH will have a pH that is

## **Assessment Chapter Test B - marshallh. weebly.com**

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Chemistry Chapter 7:  
Chemical Formulas and  
Chemical Compounds  
Chapter Exam  
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Chapter Test B  
Answer Key

your answers to the questions and click 'Next' to see the next set of questions...

## **Holt McDougal Modern Chemistry Chapter 7: Chemical**

...

Modern Chemistry 110

Chapter Test Name

Class Date Chapter

Test B, continued \_\_\_\_\_

7. Effervescence is the  
a. dissolving of a gas in  
a liquid. b. escape of a  
liquid from a liquid-

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Assessment  
Chapter 7 Test B  
Answer Key

liquid solution. c.  
escape of a solid from  
a solid-liquid solution.  
d. escape of a gas from  
a gas-liquid solution.  
\_\_\_\_\_ 8.

## **Assessment Chapter Test B - Ed W. Clark High School**

Holt McDougal Modern  
Chemistry 3 Chapter  
Test Chapter Test B,  
continued 16 Modern  
chemistry chapter 3  
test b answers. The  
measure of the ability

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## Assessment

of an atom in a  
chemical compound to  
attract electrons from  
another atom in the  
compound is called  
\_\_\_\_. 17. The energy  
required to remove one  
electron from an atom  
is called its \_\_\_\_\_. 18.

### **Modern Chemistry Chapter 3 Test B Answers**

Modern Chemistry •  
CHAPTER 7  
HOMEWORK 7-8 (pp.  
224-226) SKILL  
*Page 22/28*

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**BUILDER** Explain how to find the answer to each of the following questions. The first one is done for you. 1. How many molecules are found in a certain sample of a compound?

## **Modern Chemistry Chapter 7 Homework 7 4 Answers**

CHAPTER 7 REVIEW  
Chemical Formulas and  
Chemical Compounds  
SECTION 1 SHORT

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ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron (III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom.

## **Holt Modern Chemistry Chapter 7 Review Answer Key**

chemical formula



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## Assessment Chapter 7 Test B Answer Key

subscript of each  
element reveals  
number of atoms  
contained in a single  
molecule of a certain  
compound monoatomic  
ions ions formed from  
a single atom;  
remember to change  
nonmetals to -ide  
ending oxyanions  
polyatomic ions that  
contain oxygen binary  
ionic compounds  
compounds of 2  
elements between  
metal (positive, s-block

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Assessment  
and p-block) and  
nonmetal ...  
Chapter 7 Test B

## Answer Key **Modern Chemistry - Holt Chapter 7 | StudyHippo.com**

Modern Chemistry 141

Chapter Test Name

Class Date Chapter

Test B, continued 18

When a strong acid is

titrated with a weak

base, the pH of the

solution at the

equivalence point is

than 7 19 A is a highly

purified solid used to

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## Chapter 7 Test B

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Will Make You Smile |  
Short Film Showcase -  
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27,995,446 views

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cd98f00b204e9800998

ecf8427e. Chapter 7 Test B

## Answer Key