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Modern Chemistry • CHAPTER 7. HOMEWORK 7-1. (pp. 203–206) VOCABULARY Write true or false for each statement. 1. A chemical formula indicates the relative number of molecules of each kind in a chemical compound. 2. Monoatomic ions are ions formed from a single atom. 3.

HOMEWORK 7-1

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Modern Chemistry • CHAPTER 7 HOMEWORK 7-7 VOCABULARY Complete each sentence. Use an additional sheet of paper if necessary. 1. The formula mass of a molecule or ion is _____. 2. The molar mass of an element is equal to _____. 3. To calculate the molar mass of a compound, you should _____. 4. A compound's molar mass is numerically equal to ...

HOMEWORK 7-7

Modern Chemistry • CHAPTER 7 HOMEWORK 7-8 (pp. 224–226) SKILL BUILDER Explain how to find the answer to each of the following questions. The first one is done for you. 1. How many molecules are found in a certain sample of a compound?

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Modern Chemistry • CHAPTER 7 HOMEWORK 7-3 (pp. 211–213) VOCABULARY Write the prefix that corresponds to each number. Try not to use the book. Number Prefix Number Prefix 16 27 38 49 510 GRAPHIC ORGANIZER Complete the chart by supplying the missing information. Name of Compound Formula of Compound CO₂ Nitrogen dioxide S₂ Cl₂ Carbon tetrachloride P₄O₇ SF₆ Ice

HOMEWORK 7-3

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Modern Chemistry Chapter 7. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. dvann123. Chemical Names & Formulas, Oxidation Numbers, Using Chemical Formulas, & Determining Chemical Formulas Terms. Terms in this set (32) monatomic ion. an ion formed from a single atom. (remember to change nonmetals to -ide ending)

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CHAPTER 7 • HOMEWORK 7-1 VOCABULARY 1. false 2. true 3. false 4. false 5. true SKILL BUILDER AgCl—In the compound, there is one atom of silver for each atom of chlorine. Ba(OH)₂ —The subscript 2 refers to everything inside the parentheses, so there are two hydroxide ions for every atom of barium. (NH₄)₂ SO₄

HOMEWORK 7-2

Modern Chemistry • CHAPTER 7 HOMEWORK 7-6 SKILL BUILDER Determine the oxidation numbers of these compounds and ions. N₂ N₃-N₂O₃ KNO₃ NH₄NO₃ K₂Cr₂O₇ CO₃²⁻ NH₄¹⁺ ANOTHER SKILL BUILDER Name each compound according to the prefix system. Carbon(IV) iodide Sulfur(VI) oxide

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(g) Consider the number of moles of sulfurous acid be and that of be. The total number of moles of sulfurous acid is equal to the total number of moles of in the titration.. We have concentration of NaOH is 0.0100 M and volume of NaOH needed to reach the equivalence point is 31.6 mL, and volume of sulfurous acid is 1.00 L.

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Modern Chemistry CHAPTER 7 HOMEWORK 7-5 VOCABULARY Complete each sentence. 1. Oxidation numbers are assigned to the atoms composing a compound or ion in order to .. Browse and Read Modern Chemistry Chapter 5 Homework 7 Answers Modern Chemistry Chapter 5 Homework 7 Answers Spend your few moment to read a book even only few pages..

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