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Skill Practice 30 #'s 1-5

Skill Practice 1 Name: _____ Date: _____ Hour: _____ For problems 1-3, please use these conversion factors: 1 pallet = 45 bundles 1 bundle = 32 cases 1 case = 12 cans 1 can = 218.4 mL 1. How many mL of apple juice does the company need to make to fulfill an order for 2.5 pallets? 9,434,880 mL 2. If 46,680 mL of juice are produced, how many ...

Skill Practice 1

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Skill Practice 31

How many moles of Iron (Fe) are present in a sample containing 4.41×10^{22} atoms? moles How many molecules of Oxygen (O₂) are present in a sample that is 1.83×10^{-23} moles? molecules How many moles of Hydrogen (H₂) are present in a sample weighs 8.57 g? moles

Mass and Mole Relations Exercises

over and over again, chemists refer to it as a mole (abbreviated mol). 6.022×10^{23} objects is called a mole, just as you call 12 objects a dozen. Look again at how these quantities are related. 55.847 g of iron 6.022×10^{23} iron atoms 1 mol of iron 32.066 g of sulfur 6.022×10^{23} sulfur atoms 1 mol of sulfur General Plan for Converting Mass, Amount,

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CHEMQUEST 31 USING MOLES WITH FORMULAS ANSWERS PDF

Skill Practice 30 Conversion Practice Name: _____ Date: _____ Hour: _____ 1. Calculate the number of molecules in 210 grams of 'water. 7.02×10^{23} molecules 2. If you have 6.25×10^{23} molecules of sulfur tetrafluoride, how many grams do you have?

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Skill Practice 36 Name: _____ Date: _____ Hour: _____ 1. Under water where the temperature is 17°C and the pressure is 394 kPa, a diver inhales 2.1 L of air from his SCUBA tank. a) How many moles of gas are in his lungs? 0.343 mol b) If the diver swims to the surface without exhaling where the temperature is 32°C and the

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